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Lehmann

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(54) METHOD AND SYSTEM TO SHIELD MOBILE PHONES AND DEVICES TO MINIMIZE RADIATION EXPOSURE

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- (63) Continuation-in-part of application No. 12/978,142, filed on Dec. 23, 2010.
- (60) Provisional application No. 61/290,492, filed on Dec. 28, 2009, provisional application No. 61/290,493, filed on Dec. 28, 2009, provisional application No. 61/290,854, filed on Dec. 29, 2009.
- (51) Int. Cl. H04M 1/00 (2006.01)H04M 1/23 (2006.01)(2006.01)H01Q 1/24 H01Q 1/52 (2006.01)H04B 1/3827 (2015.01)
- (52) U.S. Cl. (2013.01); H01Q 1/526 (2013.01); H04B 1/3838 (2013.01)
- Field of Classification Search USPC 455/575.5, 90.3, 117, 128, 300, 347, 455/349

See application file for complete search history.

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(Continued)

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WO	2009/130165 A2	10/2009

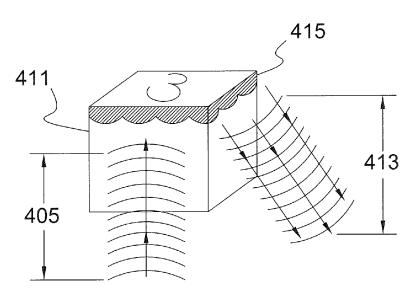
Primary Examiner — Edward Urban Assistant Examiner — Rui Hu

(74) Attorney, Agent, or Firm — Duane Morris LLP

(57)ABSTRACT

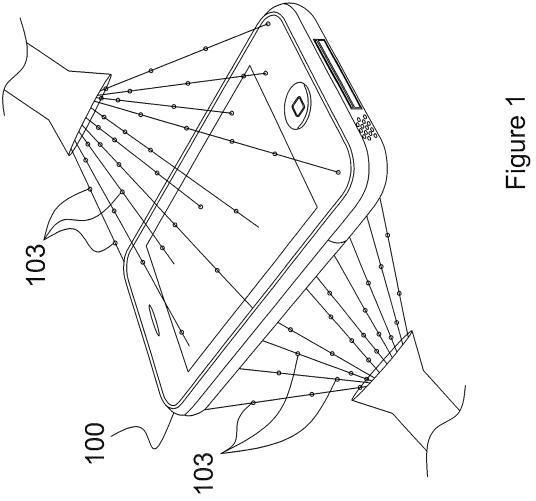
A method and system for protecting a user from exposure from radio waves emitted from a wireless transmitter include at least a layer of dielectric material to operate on the radio wave through reflection, refraction, diffraction, absorption, polarization and/or scattering of the wave. The characteristics and arrangement of multiple layers of shielding on portions of or all of a face plate may significantly reduce exposure to electromagnetic waves and the deleterious effects associated with over exposure. The shielding can be applied subsequent to manufacture, during manufacture or incorporated into structures which may form the transmitting device.

2 Claims, 9 Drawing Sheets



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(56)		Referen	ces Cited			Castaneda et al 235/145 R Lee et al 313/582
	U	S. PATENT	DOCUMENTS	2009/0124294 A1*	5/2009	Kim et al 455/566
			McDowell et al.		5/2010	Wong et al 455/575.5
			Kim et al			Wong et al
		82 * 8/2014 6/2002	Kawamura 455/11.1 Lee	* cited by examiner		,



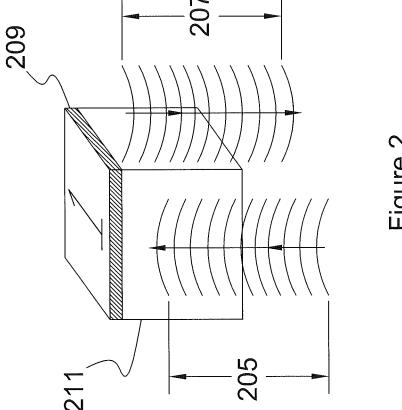
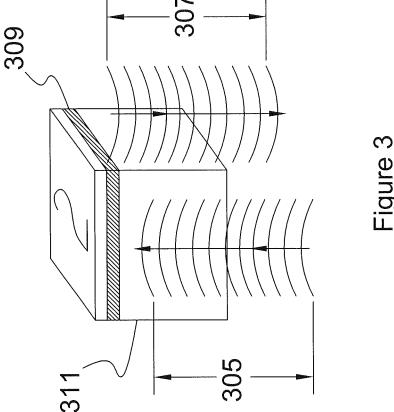


Figure 2



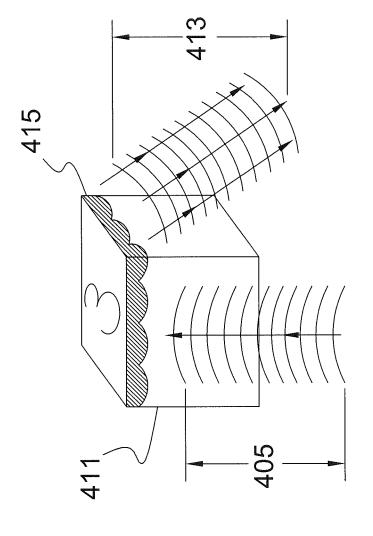


Figure 4

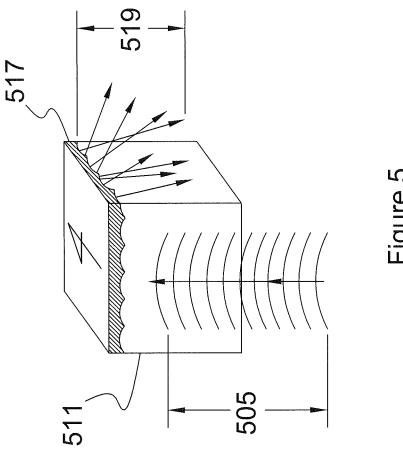


Figure 5

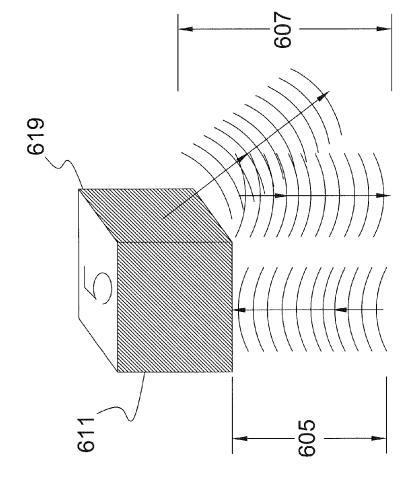


Figure 6

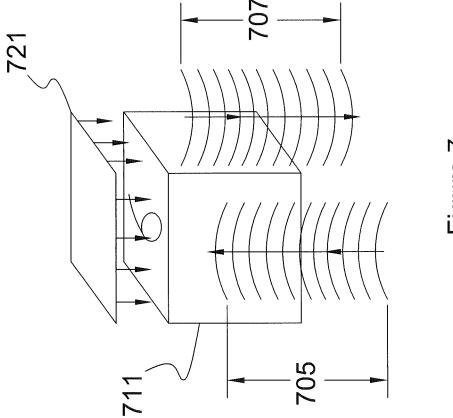
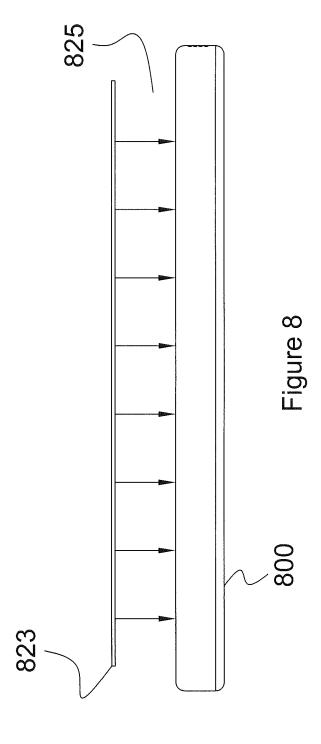
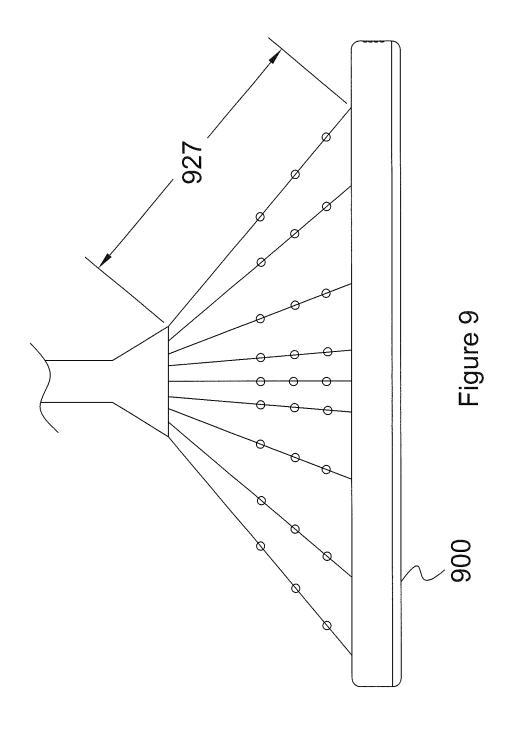


Figure 7





METHOD AND SYSTEM TO SHIELD MOBILE PHONES AND DEVICES TO MINIMIZE RADIATION EXPOSURE

CROSS REFERENCE

This application is a non-provisional application claiming priority of provisional applications: Application No. 61/290, 492 entitled METHOD AND DEVICE FOR RADIO WAVE SHIELDING THROUGH INCORPORATION OF DIELEC- 10 TRIC MATERIAL WAVE BARRIER IN ELECTRONIC DEVICE filed Dec. 28, 2009; Application No. 61/290,493 entitled METHOD AND DEVICE FOR RADIO WAVE SHIELDING THROUGH INCORPORATION OF A TRANSPARENT, SEMI-TRANSPARENT, OR OPAQUE 15 DIELECTRIC filed Dec. 28, 2009; Application No. 61/290, 854 entitled METHOD AND DEVICE FOR RADIO WAVE SHIELDING THROUGH INCORPORATION OF A TRANSPARENT, SEMI-TRANSPARENT, OR OPAQUE OR SOLID DIELECTRIC WAVE BARRIER FOR THE 20 FACEPLATE OF ELECTRONIC COMMUNICATION DEVICES, INCLUDING CELLULAR TELEPHONE, PORTABLE TELEPHONE, AND HANDHELD RADIO AND ENTERTAINMENT DEVICES filed Dec. 29, 2009. The entirety of each referenced provisional application is 25 incorporated herein by reference.

This application also claims priority benefit of non-provisional application Ser. No. 12/978,142 entitled METHOD AND SYSTEM TO MINIMIZE RADIATION EXPOSURE FROM MOBILE PHONES AND DEVICES filed Dec. 23, 30 2010. The entirety of which is also incorporated herein by reference.

BACKGROUND

The disclosed subject matter is directed to minimizing the exposure to electromagnetic waves and non-ionizing radiation due to use of mobile appliances. This disclosure uses mobile device, mobile appliance, cell phone, cellular phone, wireless phone, wireless device, mobile phone and mobile 40 radio, as well as phone, telephone, cell and radio interchangeably and the use of such terms is not exclusive and is intended to encompass all communication devices and apparatuses that transmit radio waves during communications and are held or located in close proximity to the user.

Microwaves are absorbed by living tissues at 24 times the rate of their absorption by pure water. While the Specific Absorption rate, one standard by which cellular microwave absorption by the body is commonly measured, is typically based upon a penetration through an inert emulation of a 50 human head, competent experimental work indicates that the level of absorption in living tissue is many times greater than the level of microwave absorption through an inert liquid, such as water.

In a 1983 article entitled An Optical Method for Investigating the Microwave Characteristics of DNA and other Biomolecules in Solution, by Mays L. Swicord and Christopher C. Davis, published in the journal Bioelectronmagnetics, experimental results were reported which determined that "A significant increase in the absorption of DNA solutions as compared with pure water has been observed that is consistent with microwave absorption by the longitudinal mode of the double helix". Thus, in their conclusion Swicord and Davis state that: However, the results presented in this work are in good agreement with the Prohofsky model of acoustic 65 mode absorption by varying lengths of DNA. Prohofsky and Van Zandt predicted that 450 to 2000 base pair segments of

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synthetic DNA should absorb 10³ to 10⁴ times as strongly in the microwave region as an equivalent mass of water with a decrease in peak absorption due to water damping. The 1.7% dilution of DNA investigated by PFLOH spectroscopy in this work indicated a 40% increase in absorption above pure water at 8 GHz and at 10 to 12% increase at 12 GHz. The measured DC conductance of this DNA sample was quite low yet its absorption coefficient was still 25% higher at 8 GHz than a saline solution of 20 times greater DC conductance. We conclude therefore that the observed absorption of the DNA solution does not come from ionic behavior.

The observed absorption is suggestive of direct microwave absorption by the longitudinal acoustic mode of the double helix discussed by Prohofsky and co-workers. Based on the concentration of DNA solution which gave 40% more absorption than pure water at 8 GHz, the microwave absorption of DNA at this frequency is 24 times greater than an equivalent mass of water.

This finding that there are many orders of magnitude of increase in microwave absorption when DNA is present, may likely explain the considerable body of experimental evidence strongly indicating a relationship between long term mobile device use, and diseases involving DNA strand abnormality, such as brain cancer.

Scientific [i.e, H. Lai et al, from the University of Washington, 1984, 1988, and as presented in 1998, Vienna, Austria, and 2009 in Stavanger, Norway; O. Johansson, Associate Professor, Dept. of Neuroscience of the Karolinska Institute, Stockholm, and Professor, Royal Institute of Technology, Stockholm, as presented in 2009 at Stavanger, Norway; Carl F. Blackman a founder of the Bioelectromagnetics Society, as presented in 2009, at Stavanger, Norway; Martin Blank, Ph.D., Associate Professor, Columbia University, as presented in 2009 Stavanger, Norway, Franz Adlkofer, M. D., doctorate from the Max Planck Institute for Biochemistry as presented at Stavanger Norway, 2009, also the following presenters at the International EMF Conference 2009 at Stavanger, Norway: Lukas h. Margaritis, Ph.D., Greece; Elihu D. Richter, M D, M.P.H., Israel; Alvaro Agusto A. de Salles, Ph.D., Brazil; Fredrik Soderqvist, Ph.D., Sweden, Yuri G. Grigoriev M.D. Sci., Russia; Anton V. Merkulov Ph.D., Russia] studies have shown potential health risks, in some instances showing DNA breakage, as associated with human exposure to non-ionizing radio wave sources, including but not limited to those emitted from mobile telephone devices and handsets, including but not limited to cellular telephones, satellite telephones, cordless telephones, and also portable computers, including those equipped with wi fi connectivity capacity. Continuing research supports that such health risks as may be associated with proximity to non-ionizing radiation will be reduced if level of exposure is reduced.

As used here, the term "non-ionizing radiation," is in reference to types and frequencies of radiation which do not have the capacity to force electron shifting at an atomic level, including with molecular effect. As used here, the term "ionizing radiation" is in reference to types and frequencies of radiation which do have the capacity to force electron shifting at an atomic level, including with molecular effect.

Current electromagnetic field generating communication devices allow the transmission of electromagnetic field though the human head, and are typically held in direct contact with the head of the user. While the SAR standards currently viewed as "safe," by the FCC rely upon an assumption that the device involved is held at an inch from the head of the user, the practical experience of the typical end user is to hold the device in direct contact with the ear. Microwave devices, including variants of mobile and portable telephone

devices broadcast electromagnetic waves, including but not limited to the radio waves which serve as conduit for the transmission of information, by modulation of amperage or frequency or varying combinations thereof. As discussed above there is competent scientific data indicating that information carrying microwave broadcast near the human body will result in DNA damage and can result in the formation of cancers. In addition, other data indicates that cellular device users with smaller skulls may receive a microwave dosage in inverse relationship to relative skull size, and that, further, the shape of the interior of the skull, including as to concavity, may have a focusing effect, potentially resulting in "hot spots" of electromagnetic signal absorption.

Current electromagnetic field generating communication devices, allow the transmission of electromagnetic fields through all or most quadrants of the devices involved, and specifically through the keys of such devices, which devices are, in common usage, often held near the head of the user. Current such devices, broadcast electromagnetic waves, 20 including but not limited to the radio waves which serve as conduit for information carrying by modulation of amperage or frequency or varying combinations thereof, through the keys of such devices. By the term "keys," reference is made to the actuation keys of such devices as used for telephone 25 dialing, Internet acquisition dialing, text messaging, computation, and, in short, each and all tasks which are accomplished by the users of such portable devices, and also other devices of variable portability, including but not limited to portable and stationary computers, including those with 30 "wireless," or "wi fi," capacity, all of which use keys on a keypad for the insertion of data into the device involved.

It is an object of the disclosed subject matter to reduce human user electromagnetic wave exposure from such devices through the insertion of a dielectric layer in each of 35 the keys of the keypad of such devices with insertion of such dialectic layer has a wave reflective, and therefore wave insulating characteristic, thereby working to protect the end user of such equipment from the extent of radio wave exposure to which such end user(s) would be exposed if such dielectric 40 barrier layer was not integrated into the keys of such devices. These dielectric layers once installed upon and within the keys of such devices, and by their compositional nature having a wave exposure insulating or reducing effect would reduce the level of exposure of the end user to radio waves 45 generated by the devices as described in this Application.

These and many other advantages of the present subject matter will be readily apparent to one skilled in the art to which the invention pertains from a perusal of the claims, the appended drawings, and the following detailed description of 50 preferred embodiments.

BRIEF DESCRIPTION OF THE DRAWINGS

- FIG. 1 is an illustration of an application of a dielectric 55 material according to an embodiment of the disclosed subject matter.
- FIG. 2 is an illustration of a dielectric layer at the top of the key of a mobile device according to an embodiment of the disclosed subject matter.
- FIG. 3 is an illustration of a dielectric layer inside and below the surface of each key according to another embodiment of the disclosed subject matter.
- FIG. 4 is an illustration of dielectric layer in a convex shape, relative to the wave source, on top of the key of a 65 mobile device according to embodiments of the disclosed subject matter.

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- FIG. 5 is an illustration of a dielectric layer in a concave shape, relative to the wave source, on top of the key of a mobile device according to embodiments of the disclosed subject matter.
- FIG. **6** is an illustration of a key of a mobile device composed substantially of a dielectric material according to embodiments of the disclosed subject matter.
- FIG. 7 is an illustration of a key to which a transparent dielectric material is applied according to embodiments of the disclosed subject matter.
- FIG. 8 is an illustration of a transparent dielectric sheet placed over the face plate of a mobile device according to embodiments of the disclosed subject matter.
- FIG. 9 is an illustration of the application of a translucent dielectric material of a carrier and suspended dielectric nanospheres to a screen of a mobile device according to an embodiment of the disclosed subject matter.

DETAILED DESCRIPTION

Embodiments of devices and methods described herein reduce end user exposure to radiation. While these devices and methods are intended for protection from non-ionizing radiation, such as that emitted from cellular telephones, each embodiment and illustration may also be practiced with regard to ionizing radiation.

FIG. 1 shows the spray on application of dielectric nanospheres 103 and other wave interfering spheres such as ferrite. The dielectric coating is shown as covering the faceplate of the mobile device 100 which includes the screen and/or the key pad. In addition, portions of the back plate that do not interfere with broadcast and reception of communication signals may also be coated with a dielectric material. The dielectric material/layer or coating serving to operating on radio waves through reflection, refraction, diffraction, absorption, polarization and/or scattering.

FIG. 2 shows the installation of the dielectric layer 209 at the top of each key 211. The dielectric layer 209 may be selected from the dielectric materials listed in the Listing of Dielectric Materials of table 1, however this as stated later herein, and specifically including ferrite bead, ferrite microbead, and ferrite nano-bead, protects the end user of key-equipped radio frequency generating devices, from the negative health effects of exposure to such radio frequencies, including but not limited to the negative health effects of increased Free Radical Production and DNA Strand Breakage, as noted by Lai, above, and others. As shown in FIG. 2 radio waves 205 emanating from the mobile device are reflected back as reflected wave 207.

FIG. 3 hereto shows the installation of the dielectric layer 309 inside and below the surface of each key 311. The dielectric material layer 309 protects the end user of key-equipped radio frequency generating devices, from the negative health effects of exposure to such radio frequencies. Radio wave 305 is shown as being reflected as wave 307.

FIG. 4 hereto shows the installation of the dielectric layer 415 in a convex shape, relative to the wave source, on top of the key 411. As a result of the convex shape, wave 405 is diffused and reflected back as wave 413. This embodiment further protects the end user from the effects of the subject radio frequencies through propagating interference patterns in said radio waves in their reflected state, with an increase in such localized interference being achieved through the increased interference resultant from increased signal crossing due to the dielectric's shape. Shapes of the dielectric layer may be oriented while the layer involved is in a liquid, semiliquid, or malleable plastic state through magnetic tagging of

nano-particles of such shapes as integrated into the applied medium and influenced in cured orientation through the application of magnetic field during the curing process.

FIG. 5 hereto shows the installation of the dielectric layer 517 on top of the key 511 in a concave shape, relative to the 5 wave source. Additionally its concave shape further protects the end user from the effects of the radio waves 505 through propagating interference patterns of said radio waves in their reflected state 519, with an increase in such localized interference being achieved through the increased interference 10 resultant from increased signal crossing due to such shape. It is additionally noted that such shapes may be oriented while the layer involved is in a liquid, semi-liquid, or malleable plastic state through magnetic tagging of nano-particles of such shapes as integrated into the applied medium and influenced in cured orientation through the application of magnetic field during the curing process.

FIG. 6 hereto shows a key 611 entirely composed of dielectric material 619, having by its inherent composition a wave reflective and dampening effect. The wave 605 is both attenuated and reflected as illustrated by reflected waves 607.

FIG. 7 hereto shows a key 711 to which transparent dielectric material 721 has been adhesively applied, or applied by other means of bonding, including but not limited to non-pb transparent dielectric materials such as Bi2O2-B2O3-BaO—25 Zno, such as described by Joon-Young Song and Se-Young Choi, of the School of New Materials Science and Engineering, Yonsei University, Seoul 120-749, South Korea, in their 2005, 2006 paper Fabrication and characterization of Pb-free transparent dielectric layer for plasma display panel, so has to have a wave reflective, and therefore from the perspective of the user, wave dampening effect, as to radiation which would otherwise pass through the keys into the body of the human end user. The wave 705 and the reflected wave 707 as a result of the transparent dielectric layer 721 is shown in FIG. 7.

FIG. 8 shows a transparent dielectric sheet 823 placed over the face plate of a mobile device 800. The transparent, translucent, or semi-transparent dielectric sheet 823 may be adhesively applied, or applied by other means of bonding, including but not limited to non-pb transparent dielectric materials 40 such as Bi2O2-B2O3-BaO—Zno, such as described above, so as to have a wave reflective, and therefore from the perspective of the user, wave dampening effect, as to radiation which would otherwise pass through the screen of the device into the body of the human end user. Any transparent dielectric material which may be so bonded, including by sheet, spray, weld, powder coat, or epoxy or other cure method to the device is contemplated.

FIG. 9 hereto shows a screen of a mobile device 900, to which translucent dielectric material 927, composed of a 50 carrier and suspended dielectric nano-spheres have been adhesively applied, or applied by other means of bonding, including but not limited to non-pb transparent dielectric materials such as $B_{12}O_2$ — B_2O_3 —BaO—Zno, such as described by Joon-Young Song and Se-Young Choi, and, as to 55 such spheres, including ferrite nano-spheres, so as to have a wave reflective nature, and therefore from the perspective of the user, wave dampening effect, as to radiation which would otherwise pass through the screen of the device into the body of the human end user.

An aspect of the disclosed subject matter includes arranging layers of materials and interfaces to shield the user from radio waves. Dielectric layers may be alternated with different dielectric layers, layers of different refractive indices or layers of different materials. Additionally in keys or coatings with multiple interfaces, each interface may define the same or different shapes to work independently or cooperatively.

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Another aspect of the disclosed subject matter is the use of transparent, translucent or semi-translucent dielectric material to allow for shielding of screens and other parts where opaque coatings would deleteriously affect functionality.

Dielectric material suitable for use in the disclosed subject matter may include but are not limited to those listed in Table 1.

TABLE 1

(Dielectric Materials)

ABS Resin, Lump 2.4-4.1 ABS Resin, Pellet 1.5-2.5 Acenaphthene (70 □ F.) 3.0 Acetal (70° F.) 3.6 Acetal Bromide 16.5 Acetal Doxime (68 F.) 3.4 Acetaldehyde (41 F.) 21.8 Acetamide (68 F.) 41 Acetamide (180
☐ F.) 59.0 Acetanilide (71
□ F.) 2.9 Acetic Acid (68 F.) 6.2 Acetic Acid (36 F.) 4.1 Acetic Anhydride (66□ F.) 21.0 Acetone (77 F.) 20.7 Acetone (127 F.) 17.7 Acetone (32 F.) 1.0159 Acetonitrile (70 F.) 37.5 Acetophenone (75° F.) 17.3 Acetoxime (24 F.) 3 Acetyl Acetone (68 F.) 23.1 Acetyl Bromide (68 F.) 16.5 Acetyl Chloride (68° F.) 15.8 Acetyl Acetone (68 F.) 25.0 Acetylene (32 F.) 1.0217 Acetylmethyl Hexyl Ketone (66° F.) 27.9 Acrylic Resin 2.7-4.5 Acteal 21.0-3.6 Air 1 Air (Dry) (68 F.) 1.000536 Alcohol, Industrial 16-31 Alkyd Resin 3.5-5 Allyl Alcohol (58 F.) 22.0 Allyl Bromide (66 F.) 7.0 Allyl Chloride (68 F.) 8.2 Allyl Iodide (66° F.) 6.1 Allyl Isothiocyanate (64 ☐ F.) 17.2 Allyl Resin (Cast) 3.6-4.5 Alumina 9-3-11.5 Alumina 4.5 Alumina China 3.1-3.9 Aluminum Bromide (212□ F.) 3.4 Aluminum Fluoride 2.2 Aluminum Hydroxide 2.2 Aluminum Oleate (63 F.) 2.4 Aluminum Phosphate 6.0 Aluminum Powder 1.6-1.8 Amber 2.8-2.9 Aminoalkyd Resin 3.9-4.2 Ammonia (-74□ F.) 25 Ammonia (-30□ F.) 22.0 Ammonia (40 ☐ F.) 18.9 Ammonia (69 LF.) 16.5 Ammonia (Gas?) (32° F.) .0072 Ammonium Bromide 7.2 Ammonium Chloride 7.0 Amyl Acetate (68 F.) 5.0 Amyl Alcohol (-180 F.) 35.5 Amyl Alcohol (68 F.) 15.8 Amyl Alcohol (140 F.) 11.2 Amyl Benzoate (68° F.) 5.1 Amyl Bromide (50 F.) 6.3 Amyl Chloride (52 ☐ F.) 6.6 Amyl Ether (60 ☐ F.) 3.1 Amyl Formate (66° F.) 5.7 Amyl Iodide (62 F.) 6.9 Amyl Nitrate (62 ☐ F.) 9.1 Amyl Thiocyanate (68 F.) 17.4 Amylamine (72° F.) 4.6

TABLE 1-continued TABLE 1-continued

(Dielectric Materials)		(Dielectric Materials)
Amylene (70° F.) 2.0		Boron Bromide (32 F.) 2.6
Amylene Bromide (58° F.) 5.6	5	Boronyl Chloride (202 F.) 5.2
•	9	
Amylenetetracarboxylate (66° F.) 4.4		Bromaceytal Bromide 12.6
Amylmercaptan (68° F.) 4.7		Bromal (70° F.) 7.6
Aniline (32° F.) 7.8		Bromine (68□ F.) 3.1
Aniline (68° F.) 7.3		Bromine (32□ F.) 1.0128
Aniline (212° F.) 5.5		Bromo-2-Ethoxypentane (76□ F.) 6.5
	4.0	
Aniline Formaldehyde Resin 3.5-3.6	10	Bromoacetyl Bromide (68 F.) 12.6
Aniline Resin 3.4-3.8		Bromoaniline (68□ F.) 13
Anisaldehyde (68° F.) 15.8		Bromoanisole (86 F.) 7.1
Anisaldoxine (145° F.) 9.2		Bromobenzene (68° F.) 5.4
Anisole (68° F.) 4.3		Bromobutylene (68□ F.) 5.8
Anitmony Trichloride 5.3		Bromobutyric Acid (68° F.) 7.2
Antimony Pentachloride (68° F.) 3.2		Bromoctadecane 3.53
	15	
Antimony Tribromide (212° F.) 20.9		Bromodecane (76□ F.) 4.4
Antimony Trichloride (166° F.) 33.0		Bromodeodecane (76□ F.) 4.1
Antimony Trichloride 5.3		Bromodocosane (130 ☐ F.) 3.1
Antimony Tricodide (347° F.) 13.9		Bromodoeoane (75□ F.) 4.07
Apatite 7.4		Bromoform (68° F.) 4.4
Argon (-376° F.) 1.5	20	Bromoheptane (76 F.) 5.3
Argra (68° F.) 1.000513	20	Bromohexadecane (76□ F.) 3.7
Arsenic Tribromide (98° F.) 9.0		Bromohexane (76° F.) 5.8
Arsenic Trichloride (150° F.) 7.0		Bromoisovalerio Acid (68 F.) 6.5
Arsenic Trichloride (70° F.) 12.4		Bromomethane (32 F.) 9.8
Arsenic Triiodide (302° F.) 7.0		Bromonapthalene (66 F.) 5.1
		1 , ,
Arsine (-148° F.) 2.5		Bromooctadecane (86□ F.) 3.5
Asbestos 3.0-4.8	25	Bromopentadecane (68□ F.) 3.9
Ash (Fly) 1.7-2.0		Bromophropionic Acid (68 F.) 11.0
Asphalt (75° F.) 2.6		Bromotoluene (68 F.) 5.1
Asphalt, Liquid 2.5-3.2		Bromotridecane (50° F.) 4.2
		Bromoundecane (15 F.) 4.7
Azoxyanisole (122° F.) 2.3		(— /
Azoxybenzene (104° F.) 5.1		Bronyl Chloride (94□ F.) 5.21
Azoxyphenitole (302° F.) 6.8	30	Butane (30° F.) 1.4
Return to top		Butanol (1) (68 F.) 17.8
В		Butanone (68 F.) 18.9
		Butycic Anhydride (20 F.) 12.0
D-1-17- 2.5.5.0		
Bakelite 3.5-5.0		Butyl Chloral (64□ F.) 10.0
Ballast 5.4-5.6		Butyl Chloride (68□ F.) 9.6
Ballmill Feed (Cement) 4.5	35	Butyl Oleate (77□ F.) 4.0
Balm, Refuse 3.1	33	Butyl Stearate (80 ☐ F.) 3.1
Barium Chloride 9.4		Butylacetate (66 F.) 5.1
Barium Chloride (Anbyd) 11.0		Butylamine (70 F.) 5.4
Barium Chloride (2h20) 9.4		Butyraldehyde (79□ F.) 13.4
Barium Nitrate 5.8		Butyric Acid (68° F.) 3.0
Barium Sulfate (60□ F.) 11.4		Butyric Anhydride (68□ F.) 12.0
Barley Flour 3.0-4.0	40	Butyronitrile (70□ F.) 20.7
Barley Powder 3.0-4.0		Return to top
		•
Beeswax 2.7-3.0		С
Benzal Chloride (68□ F.) 6.9		
Benzaldehyde (68□ F.) 17.8		Cable Oil (80 ☐ F.) 2.2
Benzaldoxime (68° F.) 3.8		Cabondioxide (68 F.) 1.000921
Benzene (68 F.) 2.3	45	Calcim Fluoride 7.4
	7.7	
Benzene (275° F.) 2.1		Caloite 8.0
Benzene (700° F.) 1.0028		Calcium 3.0
Benzil (202□ F.) 13.0		Calcium Carbonate 6.1-9.1
Benzonitrile (68□ F.) 26.0		Calcium Fluoride 7.4
Benzophenone (122□ F.) 11.4		Calcium Oxide, Granule 11.8
Benzophenone (68 F.) 13.0	50	Calcium Sulfate 5.6
	50	
Benzotrichloride (68° F.) 7.4		Calcium Sulfate (H2o) 5.6
Benzoyl Chloride (70□ F.) 22.1		Calcium Superphosphate 14-15
Benzoyl Chloride (32□ F.) 23.0		Camphanedione (398□ F.) 16.0
Benzoylacetone (68 F.) 29.0		Camphene (68 F.) 2.7
Benzyl Acetate (70□ F.) 5.0		Camphene (104□ F.) 2.3
Benzyl Alcohol (68° F.) 13.0	55	Campher, Crystal 10-11
Benzyl Benzoate (68□ F.) 4.8		Camphoric Imide 4 (80 ☐ F.) 5.5
Benzyl Chloride (68□ F.) 6.4		Camphorpinacone (68□ F.) 3.6
Benzyl Cyanide (68□ F.) 18.3		Caprilic Acid (18° F.) 3.2
Benzyl Cyanide (155□ F.) 6.0		Caproic Acid (160 F.) 2.6
Benzyl Salicylate (68 F.) 4.1		Caprolactam Monomer 1.7-1.9
Benzylamine (68° F.) 4.6	CO	Caprylic Acid (65 F.) 3.2
Benzylethylamine (68° F.) 4.3	60	Carbide 5.8-7.0
Benzylmethylamine (67□ F.) 4.4		Carbide, Powder 5.8-7.0
Beryl 6.0		Carbon Black 2.5-3.0
Biphenyl 20		Carbon Dioxide (32 F.) 1.6
		Carbon Dioxide, Liquid 1.6
Biwax 2.5		
		Carbon Disulfide, Liquid 2.6
Bleaching Powder 4.5	65	, 1
	65	Carbon Disulfide, Liquid 2.6 Carbon Disulphide (68 F.) 2.6 Carbon Disulphide (180 F.) 2.2

TABLE 1-continued TABLE 1-continued

(Dielectric Materials)		(Dielectric Materials)
Carbon Tetrachloride (68□ F.) 2.2		Copper Oxide 18.1
Carnauba Wax 2.9	5	Corderite 2.5-5.4
Carvenone (68° F.) 18.4		Corn 5-10
Carvol (64 F.) 11.2		Corn (Dry Granulars) 1.8
Carvone (71 F.) 11.0		Corn, Refuse 2.3-2.6
Cascin 6.1-6.8		Corning Glass 6.5
Casein Resin 6.7		Cotton 1.3-1.4
Cassiterite 23.4	10	Cotton Seed Oil 3.1
Castor Oil (60 F.) 4.7	10	Co2 (32° F.) 1.6
Castor Oil (80° F.) 2.6		Creosol (53 F.) 10.6
Castor Oil (Bydrogenated) (80 F.) 10.3		Cresol (75□ F.) 5.0
Cedrene (76 F.) 3.2		Cresol, Liquid 9-11
Cellophane 3.2-6.4		Crotonic Nitrice (68 F.) 28.0
•		
Celluloid 3.3-11	15	Crystale 3.5-4.7
Cellulose 3.2-7.5		Curnaldehyde (59 F.) 11.0
Cellulose Acetate 3.2-7		Curnene (68 F.) 2.4
Cellulose Acetate (Molding) 3.2-7.0		Cumicaldehyde (58□ F.) 10.7
Cellulose Acetate (Sheet) 4.0-5.5		Cupric Oleate 2.8
Cellulose Acetate Butyrate 3.2-6.2		Cupric Oxide (60 ☐ F.) 18.1
Cellulose Nitrate (Proxylin) 6.4	20	Cupric Sulfate 10.3
Cement 1.5-2.1	20	Cupric Sulfate (Anhyd) 10.3
Cement (Plain) 1.5-2.1		Cupric Sulfate (5h2o) 7.8
Cement, Portland 2.5-2.6		Cyanoacetic Acid (40 F.) 33.0
Cement, Powder 5-10		Cyanoethyl Acetate (68□ F.) 19.3
Cereals (Dry) 3.0-5.0		Cyanogen (73□ F.) 2.6
Cerese Wax 2.4		Cyclohedane (20□ F.) 2.0
Cesium Iodine 5.6	25	Cyclohenanone (68□ F.) 18.2
Cetyl Iodide (68□ F.) 3.3		Cycloheptasiloxane (68 F.) 2.7
Charcoal 1.2-1.81		Cyclohexane (68 F.) 2.0
Chinaware, Hard 4-7		Cyclohexane, Liquid 18.5
Chloracetic Acid (140□ F.) 12.3		Cyclohexanecarboxylic Acid (88° F.) 2.6
Chloracetone 29.8		Cyclohexanemethanol (140 F.) 9.7
Chloral (68□ F.) 4.9	30	Cyclohexanol (77 F.) 15.0
Chlorhexanone Oxime 3	30	Cyclohexanone (68° F.) 18.2
Chlorine (−50□ F.) 2.1		Cyclohexanone Oxime (192 F.) 3.0
Chlorine (32□ F.) 2.0		Cyclohexene (68 F.) 18.3
Chlorine (32 F.) 2.0 Chlorine (142° F.) 1.5		Cyclohexylamine-5 5.3
Chlorine, Liquid 2		Cyclohexylphenol (130 F.)
Chloroacetic Acid (68 F.) 21.0	35	4.0 Cyclohexyltrifluoromethane-1 (68° F.) 11.0
Chloroacetone (68□ F.) 29.8		Cyclopentane (68□ F.) 2.0
Chlorobenzene (77□ F.) 5.6		Cymene 62 2.3
Chlorobenzene (100□ F.) 4.7		Return to top
Chlorobenzene (230□ F.) 4.1		D
Chlorobenzine, Liquid 5.5-6.3		
Chlorocyclohexane (76° F.) 7.6	40	D-Cocaine 3.1
Chloroform (32□ F.) 5.5	40	D.M.T. (Dacron Powder) 1.33
Chloroform (68□ F.) 4.8		Decahydronaphtolene (68° F.) 2.2
Chloroform (212□ F.) 3.7		Decamethylcyclopentasiloxane (68☐ F.) 2.5
Chloroheptane (71 ☐ F.) 5.5		Decamethyltetrasiloxane (68 ☐ F.) 2.4
Chlorohexanone Oxime (192□ F.) 3.0		Decanal 8.1
Chlorohydrate (68□ F.) 3.3		Decane (68 F.) 2.0
Chloromethane 4 12.6	45	Decanol (68 F.) 8.1
Chloronaphthalene (76□ F.) 5.0		Decylene (62 F.) 2.7
Chloroocrane (76 F.) 5.1		Decyne (68 F.) 2.2
Chlorophetane 5.4		Deuterium (68 F.) 1.3
Chlorotoluene (68□ F.) 4.7		Deuterium Oxide (77 F.) 78.3
Chlorotoluene, Liquid 4-4.5		Dextrin 2.2-2.4
Cholesterin 2.86	50	Diacetoxybutane (76° F.) 6.64
Cholestral (80□ F.) 2.9	30	Diallyl Sulfide (68 F.) 4.9
Chorine (170□ F.) 1.7		Diamond 5.5-10.0
Chrome, Ore 7.7-8.0		Diaphenylmethane 2.7
Chrome, Pure 12		Diaplmitin 3.5
Chromite 4.0-4.2		Dibeuzofuran (212° F.) 3.0
Chromyl Choride (68□ F.) 2.6	55	Dibenzyl Sebacate (68° F.) 4.6
Cinnamaldehyde (75° F.) 16.9		Dibenzylamine (68° F.) 3.6
Cis-3-Hexene (76□ F.) 2.1		Dibroheptane (24° F.) 5.08
Citraconic Anhydride (68□ F.) 40.3		Dibromobenzene (68° F.) 8.8
Citraconic Nitrile 27		Dibromobutane (68° F.) 5.7
Clay 1.8-2.8		Dibromoethylene (Cis-1,2) (32° F.) 7.7
Clinker (Cement) 2.7		Dibromoheptane (76° F.) 5.1
Coal Tar 2.0-3.0	60	Dibromohexane (76° F.) 5.0
Coal, Powder, Fine 2-4		Dibromomethane (50° F.) 7.8
Cocaine (68 F.) 3.1		Dibromopropane (68° F.) 4.3
Coffee Refuse 2.4-2.6		Dibromopropyl Alcohol (70° F.) 9.1
COIICO 1001U0C 4.T 4.U		
Coke 1.1-2.2		Dibutyl Phthalate (86° F.) 6.4 Dibutyl Sebacate (86° F.) 4.5
	65	Dibutyl Prinaiate (86° F.) 4.5 Dibutyl Tartrate 109 9.4

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TABLE 1-continued (Dielectric Materials)		TABLE 1-continued		
		(Dielectric Materials)		
Dichloracetic Acid (72° F.) 8.2		Docosane (122° F.) 2.0		
Dichloracetone (68° F.) 14.0	5	Dodecamethylcyclohexisloxane (68° F.) 2.6		
Dichlorobenzene (127° F.) 2.8		Dodecamethylpentasiloxane (68° F.) 2.5		
Dichloroethane (68° F.) 16.7		Dodecane (68° F.) 2.0		
Dichloroethane (1,2)(77° F.) 10.3		Dodecanol (76° F.) 6.5		
Dichloroethylene (62° F.) 4.6		Dodecyne (76° F.) 2.2		
Dichloromethane (68° F.) 9.1		Dolomite 6.8-8.0		
Dichlorostyrene (76° F.) 2.6	10	Dowtherm (70° F.) 3.4		
Dichlorotoluene (68° F.) 6.9	10	Return to top		
Dictyl Phthalate 5.1		E		
Dicyclohexyl Adipate (95° F.) 4.8				
Diebenzylamine (68° F.) 3.6		Ebonite 2.5-2.9		
Diethyl Benzalmalonate (32° F.) 8.0		Emery Sand 16.5		
Diethyl Disulfide (66° F.) 15.9		Epichlorchydrin (68° F.) 22.9		
Diethyl Dl-Malate (64° F.) 10.2	15	Epoxy Resin (Cast) 3.6		
		1 2 7		
Diethyl Glutarate (86° F.) 6.7		Ethanediamine (68° F.) 14.2		
Diethyl I-Malate 9.5		Ethanethiol (58° F.) 6.9		
Diethyl Kerone (58° F.) 17.3		Ethanethiolic Acid (68° F.) 13.0		
Diethyl L-Malate (68° F.) 9.5		Ethanol (77° F.) 24.3		
Diethyl Malonate (70° F.) 7.9	20	Ethelene Diamine (18° F.) 16.0		
Diethyl Oxalate (70° F.) 8.2	20	Ethelene Oxide-1 13.9		
Diethyl Oxaloacetate (66° F.) 6.1		Ethoxy-3-Methylbutane (68° F.) 4.0		
Diethyl Racemate (68° F.) 4.5		Ethoxybenzene (68° F.) 4.2		
Diethyl Sebacate (86° F.) 5.0		Ethoxyethyl Acetate (86° F.) 7.6		
Diethyl Succinate (86° F.) 6.6		Ethoxynaphthalone (66° F.) 3.3		
Diethyl Succinosuccinate (66° F.) 2.5		Ethoxypentane (73° F.) 3.6		
Diethyl Sulfide (68° F.) 7.2	25	Ethoxytoluene (68° F.) 3.9		
Diethyl Sulfite (68° F.) 15.9	20	Ethyl Acetate (77° F.) 6.0		
Diethyl Tartrate (68° F.) 4.5		Ethyl Acetoacetate (71° F.) 15.9		
Diethyl Zinc (68° F.) 2.6		Ethyl Acetoncoxalate (66° F.) 16.1		
Diethyl 1-Malate (68° F.) 9.5		Ethyl Acetophenoneoxalate (66° F.) 3.3		
Diethyl-Dimalate 10.2		Ethyl Alcohol (77° F.) 24.3		
Diethylamine (68° F.) 3.7	30	Ethyl Alcohol (See Ethanol)		
Diethylaniline (66° F.) 5.5		Ethyl Amyl Ether (68° F.) 4.0		
Dihydrocaroone (66° F.) 8.7		Ethyl Benzene (68° F.) 2.5		
Dihydrocarvone (66° F.) 8.5		Ethyl Benzoate (68° F.) 6.0		
Diimylamine (64° F.) 2.5		Ethyl Benzoylacetate (68° F.) 12.8		
Diioamylene (62° F.) 2.4		Ethyl Benzoylacetoacetate (70° F.) 8.6		
Diiodoethylene 1 (80° F.) 4.0		Ethyl Benzyl Ether (68° F.) 3.8		
Diiodomethane (77° F.) 5.3	35	Ethyl Bromide (64° F.) 4.9		
Diisoamyl (62° F.) 2.0		Ethyl Bromoisobutyrate (68° F.) 7.9		
Diisoamylene 2.4		Ethyl Bromopropionate (68° F.) 9.4		
Diisobutylamine (71° F.) 2.7		Ethyl Butyrate (66° F.) 5.1		
Dimethoxybenzene (73° F.) 4.5		Ethyl Carbonate (68° F.) 3.1		
Dimethyl Ethyl (68° F.) 11.7	40	Ethyl Carbonate (121° F.) 14.2		
Dimethyl Ethyl Carbinol (68° F.) 11.7	40	Ethyl Cellulose 2.8-3.9		
Dimethyl Malonate (68° F.) 10.4		Ethyl Chloracetate (68° F.) 11.6		
Dimehyl Oxalate (68° F.) 3.0		Ethyl Chloroformate (68° F.) 11.3		
Dimethyl Pentane (20° F.) 1.912		Ethyl Chloropropionate (68° F.) 10.1		
Dimethyl Phthalate (75° F.) 8.5		Ethyl Cinnamate (66° F.) 5.3		
Dimethyl Sulfate (68° F.) 55.0		Ethyl Cyanoacetate (68° F.) 27.0		
Dimethyl Sulfide (68° F.) 6.3	45	Ethyl Cyclobutane (68° F.) 2.0		
Dimethyl-1-Hydroxybenzene (62° F.) 4.8		Ethyl Dodecanoate (68° F.) 3.4		
Dimethyl-2-Hexane (68° F.) 2.4		Ethyl Ether (–148° F.) 8.1		
Dimethylamine (32° F.) 6.3		Ethyl Ether (-40° F.) 5.7		
Dimethylaniline (68° F.) 4.4		Ethyl Ether (68° F.) 4.3		
Dimethylbromoethylene (68° F.) 6.7		Ethyl Ethoxybenzoate (70° F.) 7.1		
Dimethylheptane (68° F.) 1.9	50	Ethyl Formate (77° F.) 7.1		
	50	• • • • • • • • • • • • • • • • • • • •		
Dimethylpentane (68° F.) 1.9		Ethyl Formylphenylacetate (68° F.) 3.0		
Dimethylquinoxaline (76° F.) 2.3		Ethyl Fumarate (73° F.) 6.5		
Dimethyltouidine (68° F.) 3.3		Ethyl Hydroxy-Tetracarboxylate 5.9		
Dinitrogen Oxide (32° F.) 1.6		Ethyl Hydroxy-Tetrocarboxylate 2.7		
Dinitrogen Tetroxide (58° F.) 2.5		Ethyl Hydroxymethylenephenylacet 5.00		
Dioctyl Phthalate (76° F.) 5.1	55	Ethyl Hydroxymethylenomalonate 6.6		
Dioxane 1,4 (77° F.) 2.2	55	Ethyl Iodide (68° F.) 7.4		
Dipalmitin (161° F.) 3.5		Ethyl Isothiocyanate (68° F.) 19.7		
Dipentene (68° F.) 2.3		Ethyl Levulinete (70° F.) 12.1		
Dipenylamine (125° F.) 3.3		Ethyl Maleate (73° F.) 8.5		
Diphemylethane (230° F.) 2.4		Ethyl Mercaptan (68° F.) 8.0		
Diphemylothene (62° F.) 12.6		Ethyl Nitrate (68° F.) 19.7		
Diphenyl 1(66° F.) 2.5	60	Ethyl Oleate (80° F.) 3.2		
Diphenyl Ether (82° F.) 3.9		Ethyl Palmitate (68° F.) 3.2		
Diphenylamine (124° F.) 3.3		Ethyl Phenylacetate (70° F.) 5.4		
Diphenylethane (110° F.) 2.38		Ethyl Propionate (68° F.) 5.7		
Diphenytmethane (62° F.) 2.6		Ethyl Salicylate (70° F.) 8.6		
		Total CONT ((CONT) 4.1		
Dipropyl Ketone (62° F.) 12.6		Ethyl Silicate (68° F.) 4.1		
Dipropyl Ketone (62° F.) 12.6 Dipropylamine (70° F.) 2.9	65	Ethyl Stincate (68° F.) 4.1 Ethyl Stearate (104° F.) 3.0		

(Dielectric Materials)		(Dielectric Materials)
*		
Ethyl Trichloracetate (68° F.) 7.8	5	Halowax 4.5 Heavy Oil 3
Ethyl Undecanoate (68° F.) 3.6 Ethyl Valerate (68° F.) 4.7	3	Heavy Oil, C 2.6
Ethyl 1-Brobutyrate (68° F.) 8.0		Helium-3 (58° F.) 1.055
Ethyl 2-Iodopropionate (68° F.) 8.8		Helium, Liquid 1.05
Ethylamine (70° F.) 6.3		Heptadecanone (140° F.) 5.3
Ethylaniline (68° F.) 5.9		Heptane (68° F.) 1.9
Ethylbenzene (76° F.) 3.0	10	Heptane, Liquid 1.9-2.0
Ethylene Chloride (68° F.) 10.5	10	Heptanoic Acid 2.5
Ethylene Chlorohydrin (77° F.) 26.0		Heptanoic Acield (71° F.) 2.59
Ethylene Cyanide (136° F.) 58.3		Heptanone (68° F.) 11.9
Ethylene Diamine (64° F.) 16.0		Heptaonic Acid (160° F.) 2.6
Ethylene Gylcol (68° F.) 37.0		Heptyl Alcohol (70° F.) 6.7
Ethylene Iodide 3.4	15	Hexamethyldisiloxane (68° F.) 2.2
Ethylene Oxide 25 14.0		Hexane (-130° F.) 2.0
Ethylene Tetraflouride 1.9-2.0		Hexanol (77° F.) 13.3
Ethylenechlorohydrin (75° F.) 25.0		Hexanone (59° F.) 14.6
Ethylenediamine (64° F.) 16.0		Hexdecamethylcycloheptasiloxane (68° F.) 2.7
Ethylic Resin 2.2-2.3		Hexyl Iodide (68° F.) 6.6
Ethylpentane (68° F.) 1.9 Ethyltoluene (76° F.) 2.2	20	Hexylene (62° F.) 2.0 Hexyliodide (68° F.) 6.6
Ethylloluene (76° F.) 2.2 Etibine (–58° F.) 2.5		Hydrazine (68° F.) 52.0
Eugenol (64° F.) 6.1		Hydrochloric Acid (68° F.) 4.60
Return to top		Hydrocyanic Acid (70° F.) 2.3
F		Hydrocyanic Acid (32° F.) 158.0
		Hydrogen (440° F.) 1.23
Fab (From Box, 8% Moisture) 1.3	25	Hydrogen (212° F.) 1.000284
Fenchone (68° F.) 12.0		Hydrogen Iodide (72° F.) 2.9
Fermanium Tetrachloride (76° F.) 2.4		Hydrogen Bromide (24° F.) 3.8
Ferric Oleate (68° F.) 2.6		Hydrogen Bromide (-120° F.) 7.0
Ferrochromium 1.5-1.8		Hydrogen Chloride (82° F.) 4.6
Ferromanganese 5.0-5.2		Hydrogen Chloride (-188° F.) 12.0
Ferrous Oxide (60° F.) 14.2	30	Hydrogen Cyanide (70° F.) 95.4
Ferrous Sulfate (58° F.) 14.2		Hydrogen Fluoride (32° F.) 84.2
Flour 2.5-3.0		Hydrogen Fluoride (-100° F.) 17
Flourine (-332° F.) 1.5		Hydrogen Iodide (72° F.) 2.9
Flourspar 6.8		Hydrogen Peroxide (32° F.) 84.2
Fluorotoluene (86° F.) 4.2		Hydrogen Peroxide 100% 70.7
Fly Ash 1.9-2.6 Formalin 2.3	35	Hydrogen Peroxide 35% 121.0 Hydrogen Sulfide (–84° F.) 9.3
Formanin 2.3 Formamide (68° F.) 84.0		Hydrogen Sulfide (-84° F.) 9.3 Hydrogen Sulfide (48° F.) 5.8
Formic Acid (60° F.) 58.0		Hydrofluoric Acid (32° F.) 83.6
Forsterite 6.2		Hydroxy-4-Methy-2-Pentanone (76° F.) 18.2
Freon 11 (70° F.) 3.1		Hydroxymethylene Camphor (86° F.) 5.2
Freon 113 (70° F.) 2.6		Hydroxymethylenehydroxymethyleneacetoacetate 7.8
Freon 12 (70° F.) 2.4	40	Hydroxymethylenebenzyl Cyanide (68° F.) 6.0
Fuller's Earth 1.8-2.2		Hydrozine (68° F.) 52.9
Furan (77° F.) 3.0		Return to top
Furfural (68° F.) 42.0		I
Furfuraldehyde (68° F.) 41.9		
Return to top		Ido-Iodohexadeoane (68° F.) 3.5
G	45	Idoheptane (71° F.) 4.9
	<u> </u>	Idohexane (68° F.) 5.4
Gasoline (70° F.) 2.0		Idomethane (68° F.) 7.0
Gerber Oatmeal (In Box) 1.5		Idopoctane (76° F.) 4.6
Germanium Tetrachloride (77° F.) 2.4		Idotoluene (68° F.) 6.1
Glass 3.7-10		Ilmenite 6.0-7.0
Glass (Silica) 3.8	50	Inadol (140° F.) 7.8
Glass, Bead 3.1 Glass, Granule 6-7		Indonol (60° F.) 7.8
Glass, Granule 6-7 Glass, Raw Material 2.0-2.5		Iodine (107° F.) 118.0 Iodine 11
Glucoheptitol (248° F.) 27.0		Iodine (250° F.) 118.0
Glycerin, Liquid 47-68		Iodine (Granular) 4.0
Glycerol (77° F.) 42.5		Iodioctane 4.6
Glycerol (32° F.) 47.2	55	Iodioctane (24° F.) 4.62
Glycerol Phthalate (Cast Alkyd) 3.7-4.0		Iodobenzene (68° F.) 4.6
Glycol Triocetate (70° F.) 6.0		Iodoheptane (22° F.) 4.92
Glycol(77° F.) 37.0		Iodohexane (20° F.) 5.37
Glycol (122° F.) 35.6		Iodomethane (20° F.) 7.0
Glycolic Nitrile (68° F.) 27.0		Iodotolune (20° F.) 6.1
Grain 3-8	60	Iron Oxide 14.2
Graphite 12-15		Iso Butyl Alcohol 18.7-31.7
Guaiacol 0 11.0		Iso Butyl Iodide 5.8
Gypsum 2.5-6.0		Iso Butyl Nitrate 11.9
Return to top		Iso Butylamine 4.5
<u>H</u>		Iso Butyric Acid 2.7
	65	Iso Butyronitrile 20.8

TABLE 1-continued		TABLE 1-continued		
(Dielectric Materials)		(Dielectric Materials)		
Iso-Butyl Alcohol (–112° F.) 31.7		Lad Oxide 25.9		
Iso-Butyl Alcohol (32° F.) 20.5	5	Lead Acetate 2.5		
Iso-Butyl Alcohol (68° F.) 18.7		Lead Carbonate (60° F.) 18.1		
Iso-Butyl Iodide (68° F.) 5.8		Lead Chloride 4.2		
Iso-Butyl Nitrate (66° F.) 11.9		Lead Nitrate 37.7		
Iso-Butylacetate (68° F.) 5.6		Lead Nomoxide (60° F.) 25.9		
Iso-Butylamine (70° F.) 4.5		Lead Oleate (64° F.) 3.2		
Iso-Butyric Acid (68° F.) 2.7	10	Lead Oxide 25.9		
Iso-Butyronitrile 23.9-20.8	10	Lead Sulfate 14.3		
Iso-Butyronitrile (75° F.) 20.8		Lead Sulfite 17.9		
Iso-Iodohexadecane 3.5		Lead Tetrachloride (68° F.) 2.8		
Iso-Propyl Alcohol (68° F.) 18.3		Lime 2.2-2.5		
Iso-Propyl Nitrate (66° F.) 11.5		Limonene (68° F.) 2.3		
Iso-Valeric Acid (68° F.) 2.7		Linde 5a Molecular Sieve, Dry 1.8		
Isoamyl Valerate (19° F.) 3.6	15	Linoleic Acid (32° F.) 2.6-2.9		
Isoamyl Acetate (68° F.) 5.6		Linseed Oil 3.2-3.5		
Isoamyl Alcohol (74° F.) 15.3		Liquified Air 1.5		
Isoamyl Bromide (76° F.) 6.1		Liquified Hydrogen 1.2		
Isoamyl Butyrate (68° F.) 3.9		Lityium Chloride 11.1		
Isoamyl Chloracetate (68° F.) 7.8	20	Lonone (65° F.) 10.0		
Isoamyl Chloride (64° F.) 6.4	20	LPG 1.6-1.9		
Isoamyl Chloroacetate 7.8		Return to top		
Isoamyl Chloroformate (68° F.) 7.8		M		
Isoamyl Lodide (65° F.) 5.6				
Isoamyl Propionate (68° F.) 4.2		M-Bromoaniline (66° F.) 13.0		
Isoamyl Salicylate (68° F.) 5.4		M-Bromotoluene (137° F.) 5.4		
Isoamyl Valerate (66° F.) 3.6	25	M-Chloroanaline (66° F.) 13.4		
Isoamylpropionate 4.2		M-Chlorotoluene (68° F.) 5.6		
Isobuthyl Resin 1.4-2.1		M-Creosol 5		
Isobutyl Acetate (68° F.) 5.6		P-Cresol (24° F.) 5.0		
Isobutyl Alcohol (68° F.) 18.7		O-Cresol (77° F.) 11.5		
Isobutyl Benzoate (68° F.) 5.9		M-Dichlorobenzene (77° F.) 5.0		
Isobutyl Bromide (20° F.) 4.0	30	M-Dinitro Benzene (68° F.) 2.8		
Isobutyl Bromide (68° F.) 6.6	30	M-Nitrotoluene (68° F.) 23.8		
Isobutyl Butyrate (68° F.) 4.0		M-Sylene 2.4		
Isobutyl Chloride (68° F.) 7.1		M-Toluidine (64° F.) 6.0		
Isobutyl Chloroformate (68° F.) 9.2		M-Xylene (68° F.) 2.4		
Isobutyl Cyanide (74° F.) 13.3		Maganese Dioxide 5-5.2		
Isobutyl Formate (66° F.) 6.5		Magnesium Oxide 9.7		
Isobutyl Iodide (68° F.) 5.8	35	Magnesium Sulfate 8.2		
Isobutyl Nitrate (66° F.) 11.9		Malachite 7.2		
Isobutyl Rininoleate (70° F.) 4.7		Maleic Anhydride (140° F.) 51.0		
Isobutyl Valerate (66° F.) 3.8		Malolic Anhydride 51		
Isobutylamine (70° F.) 4.5		Malonic Nitrile (97° F.) 47.0		
Isobutylbenzene (62° F.) 2.3		Mandelic Nitrite (73° F.) 18.1		
Isobutylbenzoate (68° F.) 5.9	40	Mandelitrile (73° F.) 17.0		
Isobutylene Bromide (68° F.) 4.0		Mannitol (71° F.) 3.0		
Isobutyric Acid (68° F.) 2.6		Margarine, Liquid 2.8-3.2		
Isobutyric Acid (122° F.) 2.7		Melamine Formaldehyde (MF)		
Isobutyric Anhydride (68° F.) 13.9		(MF) Molding Resin 5.5-6.0		
Isobutyronitrile (77° F.) 20.8	45	(MF) With Alpha Cellulose Filler 7.2-8.2		
Isocapronitrile (68° F.) 15.7	45	(MF) With Asbestos Filler 6.1-6.7		
Isooctane 2.1-2.3		(MF) With Cellulose Filler 4.7-7.0		
Isophthalic Acid 1.4		(MF) With Flock Filter 5.0-6.0		
Isoprene (77° F.) 2.1		(MF) With Macerated Fabric Fille 6.5-6.9		
Isopropyl Alcohol 18.3		Melamine Resin 4.7-10.9		
Isopropyl Benzene (68° F.) 2.4		Menthol (42° F.) 3.95		
Isopropyl Nitrate 11.5	50	Menthol (107° F.) 4.0		
Isopropylamine (68° F.) 5.5		Menthonol (43° F.) 2.1		
Isopropylether (77° F.) 3.9		Menthonol (110° F.) 2.1		
Isoquinoline (76° F.) 10.7		Mercuric Chloride 3.2		
Isosafrol (70° F.) 3.4		Mercurous Chloride 9.4		
Return to top		Mercury (298° F.) 1.00074		
J	55	Mercury Chloride 7-14		
		Mercury Diethyl (68° F.) 2.3		
Jet Fuel (Jp4) (70° F.) 1.7		Mesityl Oxide (68° F.) 15.4		
Jet Fuel (Military Jp4) 1.7		Mesitylene (68° F.) 2.4		
Return to top		Mesitylene 3.4		
K		Methal Cyanoacetate (69° F.) 29.4		
		Methallmine (77° F.) 9.4		
Kent Wax 6.5-7.5	60	Methane (-280° F.) 1.7		
Kerosene (70° F.) 1.8		Methane, Liquid 1.7		
Kynar 2.0		Methanol (77° F.) 32.6		
Return to top		Methere Idide 5.1		
L		Methoxylangene (76° F.) 4.3		
		Methoxybenzene (76° F.) 4.3		
T 1 1 1 1 1 (C10 E) 22 0				
Lactic Acid (61° F.) 22.0 Lactronitrile (68° F.) 38.4	65	Methoxyethyl Stearate (140° F.) 3.4 Methoxyphenol (82° F.) 11.0		

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TABLE 1-continued		TABLE 1-continued
(Dielectric Materials)		(Dielectric Materials)
Methoxytoluene (68° F.) 3.5		Neoprene 6-9
Methyl Acetate (77° F.) 6.7	5	Nitric Acid (14° C.) 50.0 +/- 10.0
Methyl Acetophenoneoxalate (64° F.) 2.8	_	Nitroanisole (68° F.) 24.0
• •		. ,
Methyl Alcohol (-112° F.) 56.6		Nitrobenzal Doxime (248° F.) 48.1
Methyl Alcohol (32° F.) 37.5		Nitrobenzene (68° F.) 35.7
Methyl Alcohol (68° F.) 33.1		Nitrobenzene (77° F.) 34.8
Methyl Benzoate (68° F.) 6.6		Nitrobenzene (176° F.) 26.3
Methyl Butane (68° F.) 1.8	10	Nitrobenzyl Alcohol (68° F.) 22.0
Methyl Butyl Ketone (62° F.) 12.4	10	Nitrocellulose 6.2-7.5
Methyl Butyrate (68° F.) 5.6		Nitroethane (68° F.) 19.7
		Nitrogen (336° F.) 1.454
Methyl Chloride (77° F.) 12.9		
Methyl Chloroacetate (68° F.) 12.9		Nitrogen (68° F.) 1.000580
Methyl Ether (78° F.) 5.0		Nitroglycerin (68° F.) 19.0
Methyl Ethyl Ketone (72° F.) 18.4	15	Nitromethane 22.7-39.4
Methyl Ethyl Ketoxime (68° F.) 3.4	13	Nitromethane (68° F.) 39.4
Methyl Formate (68° F.) 8.5		Nitrosodimethylamine (68° F.) 54.0
Methyl Heptanol (68° F.) 5.3		Nitrosyl Bromide (4° F.) 13.0
Methyl Iodide (68° F.) 7.1		Nitrosyl Chloride (10° F.) 18.0
Methyl Kexyl Ketone (62° F.) 10.7		Nitrotoluene (68° F.) 1.96
Methyl Methacrylate (Cast) 2.7-3.2		Nitrous Oxide (32° F.) 1.6
Methyl Nitrobenzoate (80° F.) 27.0	20	Nonane (68° F.) 2.0
Methyl O-Methoxybenzoate (70° F.) 7.8		Nylon 4.0-5.0
Methyl P-Toluate (91° F.) 4.3		Nylon Resin 3.0-5.0
Methyl Propionate (66° F.) 5.4		Return to top
Methyl Propyl Ketone (58° F.) 16.8		O
Methyl Salicylate (68° F.) 9.0		
Methyl Thiocyanate (68° F.) 35.9	25	O-Bromotoluene (137° F.) 4.3
Methyl Valerate (66° F.) 4.3		O-Chlorophenol (66° F.) 8.2
Methyl 5 Ketocyclohexylene (68° F.) 24.0		O-Chlorotoluene (68° F.) 4.5
Methyl-1-Cyclopentanol (35° F.) 6.9		O-Cresol (77° F.) 11.5
Methyl-2 4-Pentandeiol (86° F.) 24.4		O-Dichlorobenzene (77° F.) 7.5
Methyl-2-Pentanone (68° F.) 13.1		O-Nitro Analine (194° F.) 34.5
Methylal (68° F.) 2.7	30	O-Nitrotoluene (68° F.) 27.4
Methylaniline (68° F.) 6.0	50	O-Toluidine (64° F.) 6.3
Methylbenzylamine (65° F.) 4.4		O-Xylene (68° F.) 2.6
Methylcyclohexanol (68° C.) 13.0		Octadecanol 3.42
Methylcyclohexanone (192° F.) 18.0		Octadecanol (136° F.) 3.4
Methylcylopentane (68° F.) 2.0		Octamethylcyclotetrasiloxane (68° F.) 2.4
Methylene Iodide (70° F.) 5.1		Octamethyltrisiloxane (68° F.) 2.3
Methyleneaceloacetate (70° F.) 7.8	35	Octane (24° F.) 1.061
Methylenemalonate (72° F.) 6.6		Octane (68° F.) 2.0
Methylenephenylacetate (68° F.) 5.0		Octanone (68° F.) 10.3
Methylether, Liquid 5		Octene (76° F.) 2.1
Methylhexane (68° F.) 1.9		Octyl Alcohol (64° F.) 3.4
Methylisocyanate (69° F.) 29.4		Octyl Iodide (68° F.) 4.9
Methyloctane (69° F.) 30.0	40	Octylene (65° F.) 4.1
Methylomine (21° F.) 10.5		Oil, Peanut (52° F.) 3.0
		/ /
Methylphenyl Hydrazin (66° F.) 7.3		Oil, Almond (68° F.) 2.8
Methylpyridine (2) (68° F.) 9.8		Oil, Cotton Seed (57° F.) 3.1
Metnoxy-Four-Methyl Phenol 11		Oil, Grapeseed (61° F.) 2.9
Mica 2.6-3.2		Oil, Lemon (70° F.) 2.3
Mica 7.0	45	Oil, Linseed 3.4
Mica (Glass Bonded) 6.9-9.2		Oil, Olive (68° F.) 3.1
Micanite 1.8-2.6		Oil, Paraffin (68° F.) 2.2-4.7
Mills (Dry Powder) 1.8		Oil, Petroleum (68° F.) 2.1
Mineral Oil (80° F.) 2.1		Oil, Pyranol (68° F.) 5.3
Monomyristin (158° F.) 6.1		Oil, Sesame (55° F.) 3.0
Monopalmitin (152° F.) 5.3	50	Oil, Sperm (68° F.) 3.2
Monostearin (170° F.) 4.9	30	Oil, Terpentine (68° F.) 2.2
Morpholine (77° F.) 7.3		Oil, Transformer (68° F.) 2.2
Return to top		Oleic Acid (68° F.) 2.5
N		Oleric Acid 2.4-2.5
		One-Dichloroethane 10.7
N-Butyl Alcohol (66° F.) 7.8		One-Diethoxyethane 3.8
N-Butyl Bromide (68° F.) 6.6	55	Opal Wax 3.1
N-Butyl Formate (-317° F.) 2.4		Organic Cold Molding Compound 6.0
N-Butyl Iodide (77° F.) 6.1		Oxygen (-315° F.) 1.51
N-Butylacetate (19° F.) 5.1		Oxygen (68° F.) 1.000494
N-Butyricaid (68° F.) 2.9		Return to top
N-Hexane (68° F.) 1.9		
IN-TIEARIE (UO F.) I.Y	60	<u>P</u>
	00	
N-Methylaniline (68° F.) 6.0		P-Bromotoluene (137° F.) 5.5
N-Methylaniline (68° F.) 6.0 N-Pentane (68° F.) 1.8		
N-Methylaniline (68° F.) 6.0 N-Pentane (68° F.) 1.8 Naphthy Ethyl Ether (67° F.) 3.2		P-Chlorophenol (130° F.) 9.5
N-Methylaniline (68° F.) 6.0 N-Pentane (68° F.) 1.8 Naphthy Ethyl Ether (67° F.) 3.2 Naphtalene (185° F.) 2.3		P-Chlorophenol (130° F.) 9.5 P-Chlorotoluene (68° F.) 6.1
N-Methylaniline (68° F.) 6.0 N-Pentane (68° F.) 1.8 Naphthy Ethyl Ether (67° F.) 3.2 Napthalene (185° F.) 2.3 Napthalene (68° F.) 2.5		P-Chlorophenol (130° F.) 9.5 P-Chlorotoluene (68° F.) 6.1 P-Cresol (70° F.) 5.6
N-Methylaniline (68° F.) 6.0 N-Pentane (68° F.) 1.8 Naphthy Ethyl Ether (67° F.) 3.2 Napthalene (185° F.) 2.3 Napthalene (68° F.) 2.5 Napthonitrile (70° F.) 6.4		P-Chlorophenol (130° F.) 9.5 P-Chlorotoluene (68° F.) 6.1 P-Cresol (70° F.) 5.6 P-Cresol (137° F.) 9.9
N-Methylaniline (68° F.) 6.0 N-Pentane (68° F.) 1.8 Naphthy Ethyl Ether (67° F.) 3.2 Napthalene (185° F.) 2.3 Napthalene (68° F.) 2.5	65	P-Chlorophenol (130° F.) 9.5 P-Chlorotoluene (68° F.) 6.1 P-Cresol (70° F.) 5.6

TABLE 1-continued TABLE 1-continued

(Dielectric Materials)		(Dielectric Materials)
P-Dichlorobenzine (68° F.) 2.86		Pine Tree Resin, Powder 1.5-1.8
P-Dichlorobenzine (120° F.) 2.4	5	Pinene (68° F.) 2.7
P-Nitro Analine (320° F.) 56.3		Piperidine (68° F.) 5.9
P-Nitrotoluene (137° F.) 22.2		Plaster 2.5-6.0
P-Toludine 3.0		Plastic Grain 65-75
P-Toluidine (130° F.) 5.0		Plastic Pellets 1.1-3.2
P-Xylene (68° F.) 2.3		Plastic Sulphur, Unground 1.5
Paint 5-8	10	Platinum Catalyst 6.5-7.5
Palmitic Acid (160° F.) 2.3	•	Poly Propylene 1.5
Paper (Dry) 2.0		Polyacetal 3.6-3.7
Paraffin 1.9-2.5		Polyacetol Resin 2.6-3.7
Paraffin Wax 2.1-2.5		Polyacrylic Ester 3.5
Paraldehyde (68° F.) 14.5		Polyamide 2.5-2.6
Paraldehyde (77° F.) 13.9		Polybutylene 2.2-2.3
Parawax 2.3	15	Polycaprolactam 2.0-2.5
Parrafin Chloride 2.0-2.3		Polycarbonate 2.9-3.0
Penanthiene (68° F.) 2.8		Polycarbonate Resin 2.9-3.0
Pentachloroethane (60° F.) 3.7		Polyester Resin 2.8-4.5
Pentadiene 1,3 (77° F.) 2.3		Polyester Resin (Flexible) 4.1-5.2
Pentane (68° F.) 1.8	20	Polyester Resin (Glass Fiber Filled) 4.0-4.5
Pentanol (77° F.) 13.9	20	Polyester Resin (Ridgid Cast) 2.8-4.1
Pentanone (2) (68° F.) 15.4		Polyether Chloride 2.9
Pentene (1) (68° F.) 2.1		Polyether Resin 2.8-8.1
Pentochlorethane 3.7		Polyether Resin, Unsaturated 2.8-5.2
Perlite 1.3-1.4		Polyethylene 2.2-2.4
Petroleum 2.0-2.2		Polyethylene, Pellet 1.5
	25	
Phenanthrene (230° F.) 2.7	23	Polymide 2.8
Phenathiene (68° F.) 2.8		Polymonochloro Pifluoroethylene 2.5
Phenathrene (110° F.) 2.72		Polypropylene 1.5
Phenetole (70° F.) 4.5		Polypropylene Powder 1.25
Phenol (118° F.) 9.9		Polypropylene, Pellet 1.5-1.8
Phenol (104° F.) 15.0		Polystyrene Resin 2.4-2.6
Phenol (50° F.) 4.3	30	Polystyrol 2.0-2.6
Phenol Ether (85° F.) 9.8		Polysulphonic Acid 2.8
Phenol Formaldehyde Resin (PFR) 4.5-5.0		Polytetra Fluoroethylene 2.0
(PFR) With Asbestos Filler 5.0-7.0		Polyvinyl Alcohol 1.9-2.0
(PFR) With Glass Fiber Filler 6.6-7.0		Polyvinyl Chloride 3.4
(PFR) With Mica Filler 4.2-5.2		Polyvinylchloride Resin 5.8-6.8
(PFR) With Mineral Filler (Cast) 9.0-15.0	35	Porcelain 5.0-7.0
(PFR) With Sisal Fiber 3.0-5.0		Porcelain With Zircon 7.1-10.5
(PFR) With Wood Flour Filler 4.0-7.0		Potassium Aluminum Sulphate 3.8
Phenol Resin 4.9		Potassium Carbonate (60° F.) 5.6
Phenol Resin, Cumulated 4.6-5.5		Potassium Chlorate 5.1
Phenoxyacetylene (76° F.) 4.8		Potassium Chloride 4.6
Phentidine (70° F.) 7.3		Potassium Chloronate 7.3
Phenyl Acetate (68° F.) 6.9	40	Potassium Iodide 5.6
Phenyl Ether (86° F.) 3.7		Potassium Nitrate 5.0
Phenyl Iso Thiocyanate (68° F.) 10.7		Potassium Sulfate 5.9
Phenyl Isocyanate (68° F.) 8.9		Potassium Chloromate 7.3
Phenyl Urethane 2.7		Potassium Chloride 5.0
Phenyl-L-Lropane (68° F.) 2.7	45	Propane (Liquid) (32° F.) 1.6
Phenyl-One-Iropane 2.7	45	Propanediol (68° F.) 32.0
Phenyl-I-Propane (68° F.) 1.7		Propanol (177° F.) 20.1
Phenylacetaldehyde (68° F.) 4.8		Propene (68° F.) 1.9
Phenylacetic (68° F.) 3.0		Propionaldehyde (62° F.) 18.9
Phenylacetonitrile (80° F.) 18.0		Propionic Acid (58° F.) 3.1
Phenylethanol (68° F.) 13.0		Propionic Anhydride (60° F.) 18.0
Phenylethyl Acetate (58° F.) 4.5	50	Propionitrile (68° F.) 27.7
Phenylethylene (77° F.) 2.4	50	Propy Butyrate (68° F.) 4.3
Phenylhydrazine (72° F.) 7.2		Propyl Acetate (68° F.) 6.3
Phenylsalicylate (122° F.) 6.3		Propyl Alcohol (68° F.) 21.8
Phosgene (32° F.) 4.7		Propyl Benzene (68° F.) 2.4
Phosphine (-76° F.) 2.5		Propyl Bromide (68° F.) 7.2
Phosphorus (93° F.) 4.1	55	Propyl Butyrate (68° F.) 4.3
Phosphorus Oxychloride (72° F.) 14.0		Propyl Chloroformate (68° F.) 11.2
Phosphorus Pentachloride (320° F.) 2.8		Propyl Ether (78° F.) 3.4
Phosphorus Tribromide 3.9		Propyl Formate (66° F.) 7.9
Phosphorus Tribromide (68° F.) 3.9		Propyl Nitrate (64° F.) 14.2
Phosphorus Trichloride (77° F.) 3.4		Propyl Propionate (68° F.) 4.7
Phosphorus, Red 4.1		Propyl Valerate (65° F.) 4.0
	60	Propylene Liquid 11.9
Phosphorus, Yellow 3.6	= =	
Phosphoryl Chloride (70° F.) 13.0		Psuedocumene (60° F.) 2.4
Phosphrous 4.1		Pulegone (68° F.) 9.5
Phtalide (166° F.) 36.0		Pulezone (66° F.) 9.7
Phthalic Acid 5.1-6.3		PVC, Powder 1.4
Phthalide (74° F.) 36.0		Pyrex 4.8
I hendride (7 F 1.) 50.0		
Pinacolin (62° F.) 12.8	65	Pyrex Glass 4.3-5.0

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TABLE 1-continued	TABLE 1-continued		
(Dielectric Materials)		(Dielectric Materials)	
Pyroceram 3.5-4.5		Stannec Chloride (72° F.) 3.2	
Pyrrole (63° F.) 7.5	5	Starch 3-5	
Return to top		Starch, Paste 1.7-1.8	
Q		Stearic Acid (160° F.) 2.3	
		Stearine 2.3	
Quartz 4.2		Steatite 5.5-7-5	
Quinoline (77° F.) 9.0		Styrene (77° F.) 2.4	
Quinoline (-292° F.) 2.6	10	Styrene (Modified) 2.4-3.8	
Return to top		Styrene (Phenylethane) (77° F.) 2.4	
R		Styrene Resin 2.3-3.4	
		Succinamide (72° F.) 2.9	
Reburned Lime 2.2		Succinic Acid (78° F.) 2.4	
Refractory (Cast) 6.7		Sucrose 3.3	
Refractory (For Casting) 1.8-2.1	15	Sucrose (Mean) 3.3	
Resorcinol 3.2	13	Sugar 3.0	
Rice (Dry) 3.5		Sugar, Granulated 1.5-2.2	
Rice Bran 1.4-2.0		Sulfur 1.6-1.7	
Rouge 1.5		Sulfur Dioxide (-4° F.) 17.6	
Rouge (Jewelers) 1.5-1.6		Sulfur Dioxide (32° F.) 15.0	
Rubber 3.0	•	Sulfur Monochloride (58° F.) 4.8	
Rubber (Chlorinated) 3.0	20	Sulfur Trioxide (64° F.) 3.1	
Rubber (Hard) 2.8		Sulfurous Oxychloride (72° F.) 9.1	
Rubber (Isomerized) 2.4-3.7		Sulfuryl Chloride (72° F.) 10.0	
Rubber Cement 2.7-2.9		Sulphur (244° F.) 3.5	
Rubber Chloride 2.1-2.7		Sulphur (450° F.) 3.5	
Rubber, Raw 2.1-2.7		Sulphur Dioxide (32° F.) 15.6	
Rubber, Sulphurized 2.5-4.6	25	Sulphur Trioxide (70° F.) 3.6	
Ruby 11.3		Sulphur, Liquid 3.5	
Rutile 6.7		Sulphur, Powder 3.6	
Return to top		Sulphuric Acid (68° F.) 84.0	
S		Sulfuric Acid (25° C.) 100.0	
		Supphuric Oxychloride (72° F.) 9.2	
Safrol (70° F.) 3.1	30	Syrup 50-80	
Salicylaldehyde (68° F.) 13.9	50	Syrup Wax 2.5-2.9	
Salt 3.0-15.0		Return to top	
Sand (Dry) 5.0		T	
Sand (Silicon Dioxide) 3-5.0			
Santowax (70° F.) 2.3		Tantalum Oxide 11.6	
Selenium 6.1-7.4	2.5	Tartaric Acid (68° F.) 6.0	
Selenium 11	35	Tartaric Acid (14° F.) 35.9	
Selenium (482° F.) 5.4		Teflon 2.0	
Selevium (249° F.) 5.4		Teflon (4f) 2.0	
Sesame 1.8-2.0		Teflon, FEP 2.1	
Shellac 2.0-3.8		Teflon, PCTFE 2.3-2.8	
Silica Aluminate 2		Teflon, PTFE 2	
Silica Sand 2.5-3.5	40	Tepineol 2.8	
Silicon 11.0-12.0		Terpinene (70° F.) 2.7	
Silicon Dioxide 4.5		Terpineol (72° F.) 2.8	
Silicon Tetrachloride (60° F.) 2.4		Tetrabromoethane (72° F.) 7.0	
Silicone Molding Compound (SMC)		Tetrachloroethylene (70° F.) 2.5	
(SMC) (Glass Fiber Filled) 3.7		Tetradecamethyltetradecamethylcycloheptasiloxan 2.7	
Silicone Oil 2.2-2.9	45	Tetradecamethylhexosiloxane (68° F.) 2.5	
Silicone Resin, Liquid 3.5-5.0	7.7	Tetradecanol (100° F.) 4.7	
Silicone Rubber 3.2-9.8		Tetraethyl Amylenetetracarboxylate 4.40	
Silicone Varnish 2.8-3.3		Tetraethyl Hexane-1-Phenyl Tetracarboxylate (66° F.) 5.9	
Silk 2.5-3.5		Tetracathyl Pentane Diphenyl Tetracarboxylate (68° F.) 2.7	
Silver Bromide 12.2		Tetraethyl Propane Tetracarboxylate (66° F.) 5.2	
Silver Chloride 11.2	50	Tetraethyl Silicate (60° F.) 4.1	
Silver Cyanide 5.6		Tetraethyl Silicate (68° F.) 4.1	
Slaked Lime, Powder 2.0-3.5		Tetrafluoroethylene 2.0	
Slate 6.0-7.5		Tetrahydro-B-Napthol (68° F.) 11.0	
Smithsonite 9.3		Tetranitromethane (68° F.) 2.2	
Soap Powders 1.2-1.7		Tetratriacontadiene (76° F.) 2.8	
Sodium Carbonate 5.3-8.4	55	Thallium Chloride 46.9	
Sodium Carbonate (Anhyd) 8.4		Thinner 3.7	
Sodium Carbonate (10h2O) 5.3		Thioacetic Acid (68° F.) 13.0	
Sodium Chloride 5.9		Thionyl Bromide (68° F.) 9.1	
Sodium Chloride (Salt) 6.1		Thionyl Chloride (68° F.) 9.3	
· /		Thiophene (60° F.) 2.8	
Sodium Cyanide 7.55		Thiophiosphoryl Chloride (70° F.) 5.8	
Sodium Cyanide 7.55	60	Thorium Oxide 10.6	
Sodium Cyanide 7.55 Sodium Dichromate 2.9 Sodium Nitrate 5.2	60		
Sodium Cyanide 7.55 Sodium Dichromate 2.9 Sodium Nitrate 5.2 Sodium Oleate (68° F.) 2.7	60	Thrichloroethylene (61° F.) 3.4	
Sodium Cyanide 7.55 Sodium Dichromate 2.9 Sodium Nitrate 5.2 Sodium Oleate (68° F.) 2.7 Sodium Perchlorate 5.4	60	Thrichloroethylene (61° F.) 3.4 Thujone (32° F.) 10.0	
Sodium Cyanide 7.55 Sodium Dichromate 2.9 Sodium Nitrate 5.2 Sodium Oleate (68° F.) 2.7 Sodium Perchlorate 5.4 Sodium Phosphate 1.6-1.9	60	Thrichloroethylene (61° F.) 3.4 Thujone (32° F.) 10.0 Tide (Loose From Box) 1.6	
Sodium Cyanide 7.55 Sodium Dichromate 2.9 Sodium Nitrate 5.2 Sodium Oleate (68° F.) 2.7 Sodium Perchlorate 5.4 Sodium Phosphate 1.6-1.9 Sodium Porchlorate 5.4	60	Thrichloroethylene (61° F.) 3.4 Thujone (32° F.) 10.0 Tide (Loose From Box) 1.6 Tin Tetrachloride (68° F.) 2.9	
Sodium Cyanide 7.55 Sodium Dichromate 2.9 Sodium Nitrate 5.2 Sodium Oleate (68° F.) 2.7 Sodium Perchlorate 5.4 Sodium Phosphate 1.6-1.9	60	Thrichloroethylene (61° F.) 3.4 Thujone (32° F.) 10.0 Tide (Loose From Box) 1.6	

(Dielectric Materials)		(Dielectric Materials)
Titanium Tetrachloride (68 Degrees F.) 2.8		Vinyl Chloride Resin, Hard 5.8-6.4
Tobacco 1.6-1.7	5	Vinyl Chloride Resin, Soft 2.8-4.0
Tobacco Dust (6% Moisture) 1.7		Vinyl Ether (68° F.) 3.9
Toluene (68° F.) 2.4		Vinyl Formal 3.0
Toluene, Liquid 2.0-2.4		Vinyllidene Chloride 3.0-4.0
Toluidine (68° F.) 6.0		Vycor Glass 3.8
Tolunitrile (73° F.) 18.8		Return to top
Tolyl Methyl Ether (68° F.) 3.5	10	W
Totane (111° F.) 5.5		
Tourmaline 6.3		Water 4-88
Trans-3-Hexene (76° F.) 2.0		Water (32° F.) 88.0
Transmission Oil (80° F.) 2.2		Water (68° F.) 80.4
Tribromopropane (68° F.) 6.4		Water (212° F.) 55.3
Tributylphosphate (86° F.) 8.0	15	Water (390° F.) 34.5
Trichlorethylene 3.4	13	Water (80° F.) 80.0
Trichloroacetic Acid (140° F.) 4.6		Water (Steam) 1.00785
Trichloroethane 7.5		Wax 2.4-6.5
Trichloroethylene (61° F.) 3.4		Wheat Flour 3.0-5.0
Trichlorololuene (70° F.) 6.9		Wheat Flour (Dry Powder) 1.6
Trichloropropane (76° F.) 2.4	20	White Mica 4.5-9.6
Trichlorotoluene (69° F.) 6.9	20	Wood, Dry 2-6
Trichloroxoluene 6.9		Wood, Pressed Board 2.0-2.6
Tricosanone(176° F.) 4.0		Wood, Wet 10-30
Tricresyl Phosphate (104° F.) 6.9		Return to top
Triethyl Aconitate (68° F.) 6.4		X
Triethyl Aluminum (68° F.) 2.9		
Triethyl Ethanetricarboxylate (66° F.) 6.5	25	Xylene (68° F.) 2.4
Triethyl Isoaconitate (68° F.) 7.2		Xylene, Liquid 2.2-2.6
Triethylamine (21° F.) 3.2		Xylenol 17
Triethylamine (77° F.) 2.4		Xylenol (62° F.) 3.9
Trifluoroactic Acid (68° F.) 39.0		Xylidine (68° F.) 5.0
Trifluorotoluene (86° F.) 9.2		Return to top
Trimethyl Borate (68° F.) 8.2	30	Y
Trimethyl-3-Heptene (68° F.) 2.2		Z
Trimethylamine (77° F.) 2.5		
Trimethylbenzene (68° F.) 2.3		Zinc Oxide 1.7-2.5
Trimethylbutane (68° F.) 1.9		Zinc Sulfide 8.2
Trimethylpentane 1.9		Zircon 12.0
Trimethylpentane (68° F.) 2.9	35	Zirconium Oxide 12.5
Trimethylsulfanilic Acid (64° F.) 89.0	33	Zirconium Silicate 5.0
Trinitrobenzene (68° F.) 2.2		Return to top
Trinitrotoluene (69° F.) 22.0		Numeric
Triolein (76° F.) 3.2		
Triphenylmethane (212° F.) 2.3		1,2-Dichloroethane (77° F.) 10.7
Tripolmitin (140° F.) 2.9	40	1-Diethoxyethane (75° F.) 3.8
Tristearin (158° F.) 2.8	40	1-Heptene (68° F.) 2.1
Turpentine (Wood) (68° F.) 2.2		1-Octanol (68° F.) 10.3
Two-Dichloroethane 10.7		2-Methyl-Propanol (77° F.) 17.7
Return to top		3 Dimethyl-2-Butanone 13.1
U		3-Chloro-1, Dihydroxyprone (68° F.) 31.0
Y 1 (600 F) 4 0		3-Dimethyl-2-Butanone (293° F.) 13.1
Undecane (68° F.) 2.0	45	
Undecanone (58° F.) 8.4		While mechanical amb adirector of the same to
Urea 5-8		While preferred embodiments of the present inver-

While preferred embodiments of the present invention have been described, it is to be understood that the embodiments described are illustrative only and that the scope of the invention is to be defined solely by the appended claims when accorded a full range of equivalence, many variations and modifications naturally occurring to those of skill in the art from a perusal hereof.

I claim:

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- 55 1. A mobile device for wireless communications comprising:
 - a transmitter and an antenna;
 - a face plate, said face plate oriented towards a user during use; said face plate further comprising a viewing screen and a plurality of user input keys;
 - a back plate disposed opposite from said face plate;
 - said antenna located proximate to a portion of the back plate; and
 - a dielectric material disposed on said face plate shielding the user from radio waves broadcast from the antenna during communications, wherein the portion of the back plate is not shielded; wherein said dielectric material is

Valeraldehyde (58° F.) 11.8
Valerie Acid (68° F.) 2.6
Valeronitrile (70° F.) 17.7
Vanadium Oxybromide (78° F.) 3.6
Vanadium Oxychloride (78° F.) 3.4
Vanadium Sulfide 3.1
Vanadium Tetrachloride (78° F.) 3.0
Vaseline 2.2-2.9
Veratrol (73° F.) 4.5
Vinyl Alcohol Resin 2.6-3.5
Vinyl Butyral 3.3-3.9
Vinyl Chloride (Acetate) 3.0-3.1
Vinyl Chloride (Flexible) 3.5-4.5
Vinyl Chloride (Ridgig) 2.8-3.0

Urea (71° F.) 3.5

Urea Resin 6.2-9.5 Urethan (121° F.) 14.2

Urethane (74° F.) 3.2

Return to top

Urethane Resin 6.5-7.1

Urea Formaldehyde (__f Degrees F.)

Urea Formaldehyde (Cellulose Filler) 6.4-6.9

disposed as a layer over a second material forming said plurality of user input keys, wherein an interface between the layer of dielectric material and the second material is convex with respect to the second material.

- **2**. A mobile device for wireless communications compris- 5 ing:
 - a transmitter and an antenna;
 - a face plate, said face plate oriented towards a user during use; said face plate further comprising a viewing screen and a plurality of user input keys;
 - a back plate disposed opposite from said face plate; said antenna located proximate to a portion of the back plate; and
 - a dielectric material disposed on said face plate shielding the user from radio waves broadcast from the antenna 15 during communications, wherein the portion of the back plate is not shielded; wherein said dielectric material is disposed as a layer over a second material forming said plurality of user input keys, wherein an interface between the layer of dielectric material and the second 20 material is concave with respect to the second material.

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